

---

Formerly Utilized Sites Remedial  
Action Program (FUSRAP)

---

**Maywood Chemical Company Superfund Site**

**ADMINISTRATIVE RECORD**

---

**Document Number**

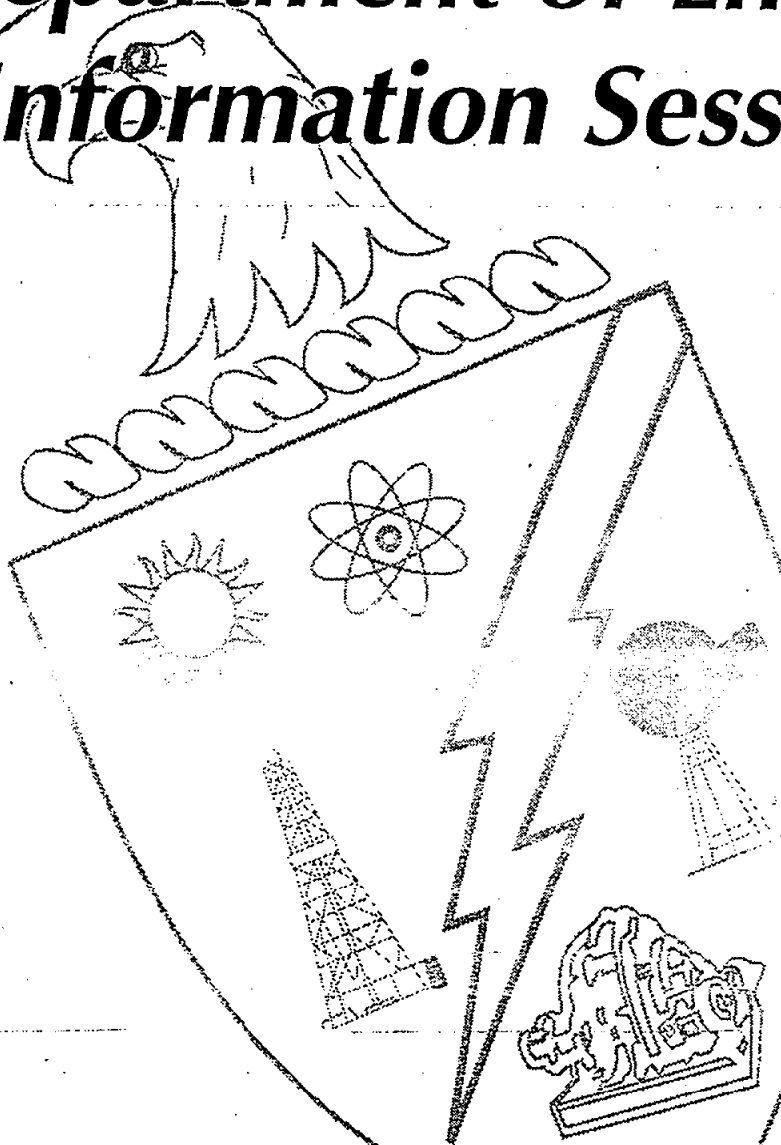
**MISS- 068.**

---



**US Army Corps  
of Engineers®**

**Department of Energy  
Information Session**



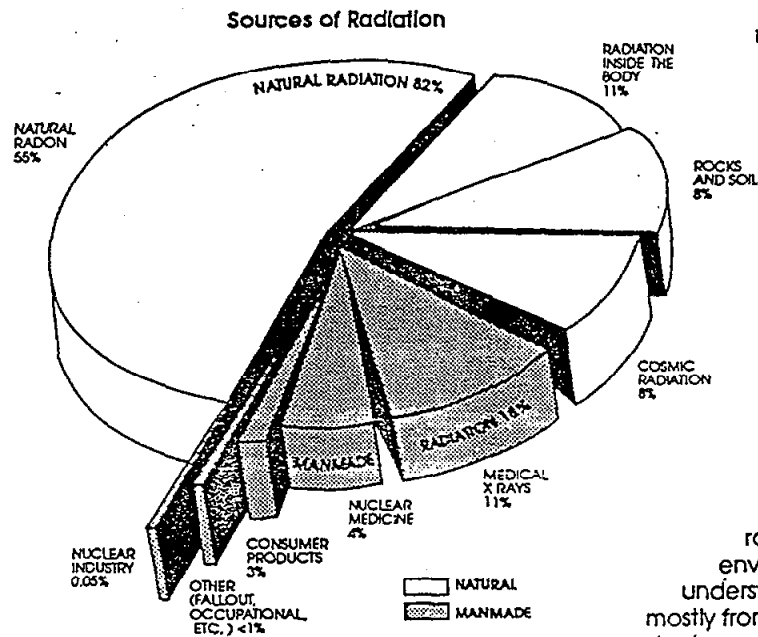
**September 28, 1993  
Maywood, New Jersey**



# Radiation in the Environment

Radiation is a natural part of our environment. When our planet was formed, radiation was present—and radiation surrounds it still. Natural radiation showers down from the distant reaches of the cosmos and continuously radiates from the rocks, soil, and water on the Earth itself.

During the last century, mankind has discovered radiation, how to use it, and how to control it. As a result, some manmade radiation has been added to the natural amounts present in our environment.



Many materials—both natural and manmade—that we come into contact with in our everyday lives are radioactive. These materials are composed of atoms that release energetic particles or waves as they change into more stable forms. These particles and waves are referred to as *radiation*, and their emission as *radioactivity*.

As the chart on the left shows, most environmental radiation (82%) is from natural sources. By far the largest source is radon, an odorless, colorless gas given off by natural radium in the Earth's crust. While radon has always been present in the environment, its significance is better understood today. Manmade radiation—mostly from medical uses and consumer products—adds about eighteen percent to our total exposure.

## TYPES OF IONIZING RADIATION

Radiation that has enough energy to disturb the electrical balance in the atoms of substances it passes through is called *ionizing radiation*. There are three basic forms of ionizing radiation.

### Alpha

Alpha particles are the largest and slowest moving type of radiation. They are easily stopped by a sheet of paper or the skin. Alpha particles can move through the air only a few inches before being stopped by air molecules. However, alpha radiation is dangerous to sensitive tissue inside the body.

### Beta

Beta particles are much smaller and faster moving than alpha particles. Beta particles pass through paper and can travel in the air for about 10 feet. However, they can be stopped by thin shielding such as a sheet of aluminum foil.

### Gamma

Gamma radiation is a type of electromagnetic wave that travels at the speed of light. It takes a thick shield of steel, lead, or concrete to stop gamma rays. X rays and cosmic rays are similar to gamma radiation. X rays are produced by manmade devices; cosmic rays reach Earth from outer space.

## Units of Measure

Radiation can be measured in a variety of ways. Typically, units of measure show either 1) the total amount of radioactivity present in a substance, or 2) the level of radiation being given off.

The radioactivity of a substance is measured in terms of the number of transformations (changes into more stable forms) per unit of time. The *curie* is the standard unit for this measurement and is based on the amount of radioactivity contained in 1 gram of radium. Numerically, 1 curie is equal to 37 billion transformations per second. The amounts of radioactivity that people normally work with are in the millicurie (one-thousandth of a curie) or microcurie (one-millionth of a curie) range. Levels of radioactivity in the environment are in the picocurie, or pCi (one-trillionth of a curie) range.

Levels of radiation are measured in various units. The level of gamma radiation in the air is measured by the *roentgen*. This is a relatively large unit, so measurements are often calculated in milliroentgens. Radiation absorbed by humans is measured in either *rad* or *rem*. The rem is the most descriptive because it measures the ability of the specific type of radiation to do damage to biological tissue. Again, typical measurements will often be in the millirem (mrem), or one-thousandth of a rem, range. In the international scientific community, absorbed dose and biological exposure are expressed in *grays* and *seiverts*. 1 gray (Gy) equals 100 rad. 1 seivert (Sv) equals 100 rem. On the average, Americans receive about 360 mrem of radiation a year. Most of this (97%) is from natural radiation and medical exposure. Specific examples of common sources of radiation are shown in the chart below.

### Cosmic Radiation

Cosmic radiation is high-energy gamma radiation that originates in outer space and filters through our atmosphere.

Sea Level .....	26 mrem/year
<small>(increases about 1/2 mrem for each additional 100 feet in elevation)</small>	
Atlanta, Georgia (1,050 feet)	31 mrem/year
Denver, Colorado (5,300 feet)	50 mrem/year
Minneapolis, Minnesota (815 feet)	30 mrem/year
Salt Lake City, Utah (4,400 feet)	46 mrem/year

### Terrestrial Radiation

Terrestrial sources are naturally radioactive elements in the soil and water such as uranium, radium, and thorium. Average levels of these elements are 1 pCi/gram of soil.

United States (average) .....	26 mrem/year
Denver, Colorado .....	63 mrem/year
Nile Delta, Egypt .....	350 mrem/year
Paris, France .....	350 mrem/year
Coast of Kerala, India .....	400 mrem/year
McAlpe, Brazil .....	2,558 mrem/year
Pocos De Caldas, Brazil .....	7,000 mrem/year

### Buildings

Many building materials, especially granite, contain naturally radioactive elements.

U.S. Capitol Building .....	85 mrem/year
Base of Statue of Liberty .....	325 mrem/year
Grand Central Station .....	525 mrem/year
The Vatican .....	800 mrem/year

### Radon

Radon levels in buildings vary, depending on geographic location, from 0.1 to 200 pCi/liter. Average Indoor Radon Level ..... 1.5 pCi/liter  
Occupational Working Limit ..... 100.0 pCi/liter

## RADIATION IN THE ENVIRONMENT

Because the radioactivity of individual samples varies, the numbers given here are approximate or represent an average. They are shown to provide a perspective for concentrations and levels of radioactivity rather than dose.

mrem = millirem  
pCi = picocurie

### Food

Food contributes an average of 20 mrem/year, mostly from potassium-40, carbon-14, hydrogen-3, radium-226, and thorium-232.

Beer .....	390 pCi/liter
Tap Water .....	20 pCi/liter
Milk .....	1,400 pCi/liter
Salad Oil .....	4,900 pCi/liter
Whiskey .....	1,200 pCi/liter
Brazil Nuts .....	14 pCi/g
Bananas .....	3 pCi/g
Flour .....	0.14 pCi/g
Peanuts & Peanut Butter .....	0.12 pCi/g
Tea .....	0.40 pCi/g

### Medical Treatment

The exposures from medical diagnosis vary widely according to the required procedure, the equipment and film used for x rays, and the skill of the operator.

Chest X Ray .....	10 mrem
Dental X Ray, Each .....	100 mrem

### Consumer Goods

Cigarettes—two packs/day (polonium-210) .....	8,000 mrem/year
Color Television .....	<1 mrem/year
Gas Lantern Mantle (thorium-232) .....	2 mrem/year
Highway Construction (cosmic) .....	0.5 mrem/hour
Airplane Travel at 39,000 feet (radon-222) .....	2 mrem/year
Phosphate Fertilizers .....	4 mrem/year

### Natural Radioactivity In Florida Phosphate Fertilizers (in pCi/gram)

	Normal Superphosphate	Concentrated Superphosphate	Gypsum
Ra-226	21.3	21.0	33.0
U-238	20.1	58.0	6.0
Th-230	18.9	48.0	13.0
Th-232	0.6	1.3	0.3

Porcelain Dentures (uranium) .....	1,500 mrem/year
Radioluminescent Clock (promethium-147) .....	<1 mrem/year
Smoke Detector (americium-241) .....	0.01 mrem/year

### International Nuclear Weapons Test Fallout from pre-1980 atmospheric tests

(average for a U.S. citizen) ..... 1 mrem/year

### References

- Effect of Ionizing Radiation on Human Health. The. Arthur C. Upton. New York University Medical Center. Atomic Industrial Forum, 1984.  
Effects on Populations of Exposure to Low Levels of Ionizing Radiation: 1980. Committee on the Biological Effects of Ionizing Radiation. National Academy Press, 1984.  
Ionizing Radiation Exposure of the Population of the United States: Report Number 93. National Council on Radiation Protection and Measurements, 1987.  
Radiation Exposure of the U.S. Population from Consumer Products and Miscellaneous Sources: Report Number 95. National Council on Radiation Protection and Measurements, 1987.  
Radiation in Medicine and Industry. A.P. Jacobson and G.P. Sokolovsky, 1980.  
Radioactivity in Consumer Products. U.S. Nuclear Regulatory Commission, 1978.

## PERSPECTIVE: How Big is a Picocurie?

The *curie* is a standard measure for the intensity of radioactivity contained in a sample of radioactive material. It was named after French scientists Marie and Pierre Curie for their landmark research into the nature of radioactivity.

The basis for the curie is the radioactivity of one gram of radium. Radium decays at a rate of about 2.2 trillion disintegrations ( $2.2 \times 10^{12}$ ) per minute. A *picocurie* is one trillionth of a curie. Thus, a picocurie represents 2.2 disintegrations per minute.

To put the relative size of one *trillionth* into perspective, consider that if the Earth were reduced to one trillionth of its diameter, the "pico earth" would be smaller in diameter than a speck of dust. In fact, it would be six times smaller than the thickness of a human hair.

The difference between the curie and the picocurie is so vast that other metric units are used between them. These are as follows:

Millicurie =	$\frac{1}{1,000}$ (one thousandth) of a curie
Microcurie =	$\frac{1}{1,000,000}$ (one millionth) of a curie
Nanocurie =	$\frac{1}{1,000,000,000}$ (one billionth) of a curie
Picocurie =	$\frac{1}{1,000,000,000,000}$ (one trillionth) of a curie

The following chart shows the relative differences between the units and gives analogies in dollars. It also gives examples of where these various amounts of radioactivity could typically be found. The number of disintegrations per minute has been rounded off for the chart.

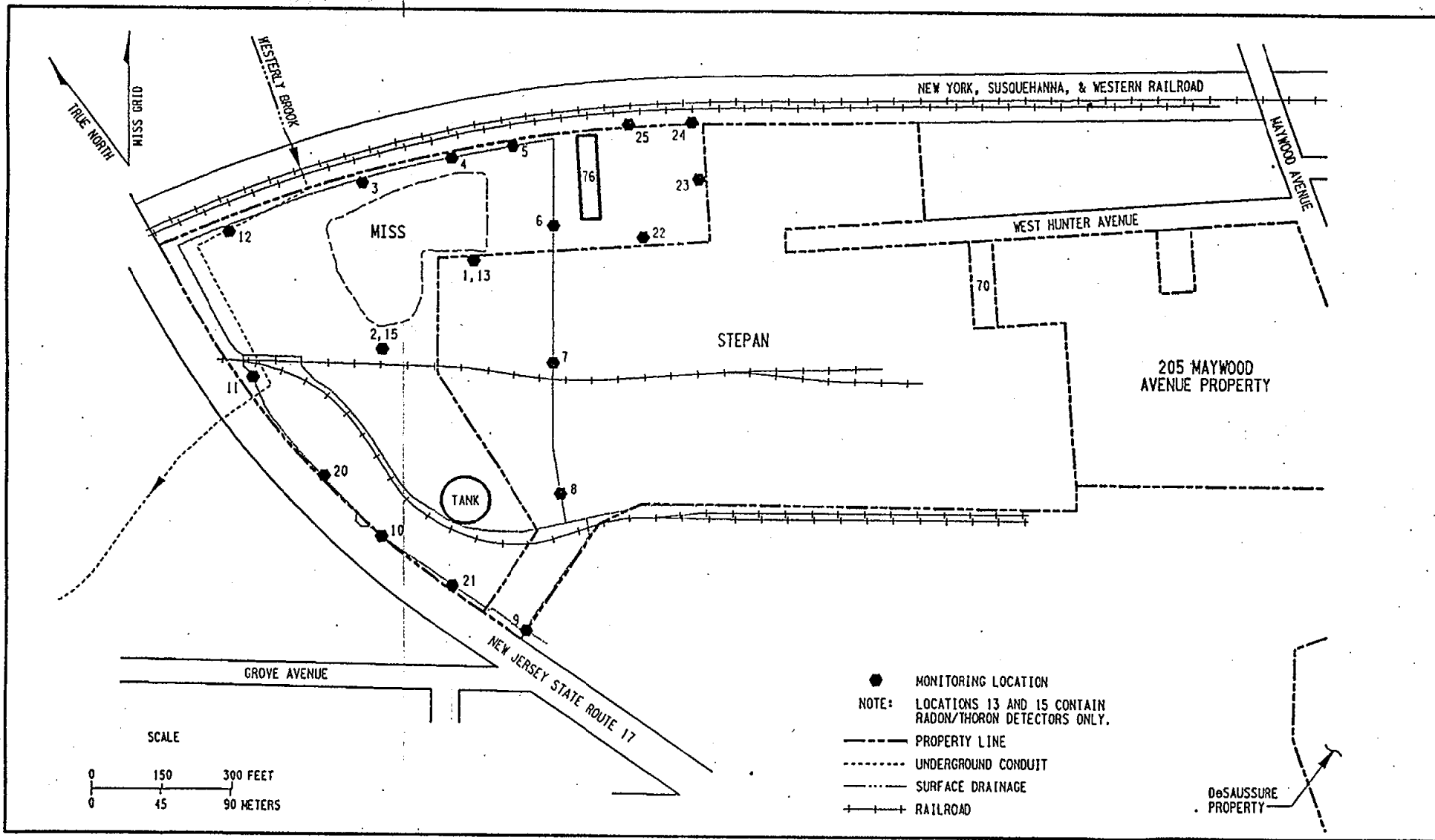
UNIT OF RADIOACTIVITY	SYMBOL	DISINTEGRATIONS PER MINUTE	DOLLAR ANALOGY	EXAMPLES OF RADIOACTIVE MATERIALS
1 Curie	Ci	$2 \times 10^{12}$ or 2 Trillion	2 Times the Annual Federal Budget	Nuclear Medicine Generator
1 Millicurie	mCi	$2 \times 10^9$ or 2 Billion	Cost of a New Interstate Highway from Atlanta to San Francisco	Amount Used for a Brain or Liver Scan
1 Microcurie	$\mu$ Ci	$2 \times 10^6$ or 2 Million	All-Star Baseball Player's Salary	Amount Used in Thyroid Tests
1 Nanocurie	nCi	$2 \times 10^3$ or 2 Thousand	Annual Home Energy Costs	Consumer Products
1 Picocurie	pCi	2	Cost of a Hamburger and Coke	Background Environmental Levels

Chart provided by W.L. Beck, Bechtel National, Inc.

# ***Comparison of 1992 Results to DOE Guidelines***

## ***Maywood Interim Storage Site***

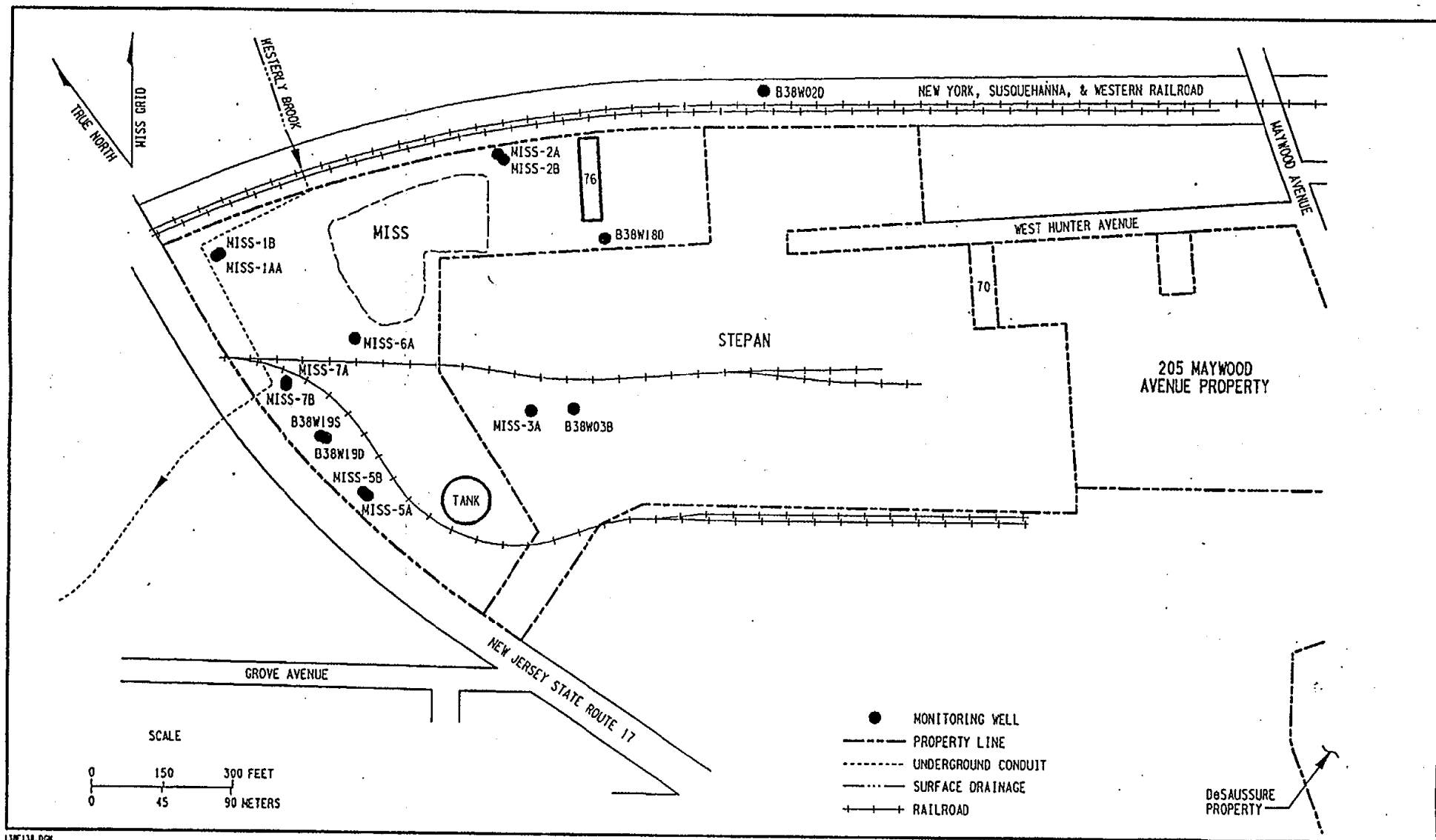
	<u><b>DOE Guidelines</b></u>	<u><b>1992 Results</b></u>	
Radon in Air	3.0 pCi/L	<0.4 pCi/L	
Thoron in Air	3.0 pCi/L	1.4 pCi/L	
External Gamma Radiation Dose (above background)	100 mrem/yr.	0.6 mrem/yr.	
		<u><b>Groundwater</b></u>	<u><b>Surface Water</b></u>
Thorium-232 in Water	50 pCi/L	1.5 pCi/L	0.2 pCi/L
Total Uranium in Water	600 pCi/L	3.3 pCi/L	0.4 pCi/L
Radium-226 in Water	100 pCi/L	0.5 pCi/L	0.3 pCi/L
		<u><b>Sediments</b></u>	
Thorium-232 in Surface Soils	5 pCi/g	1.0 pCi/g	
Radium-226 in Surface Soils	5 pCi/g	0.5 pCi/g	



130F137.DGN

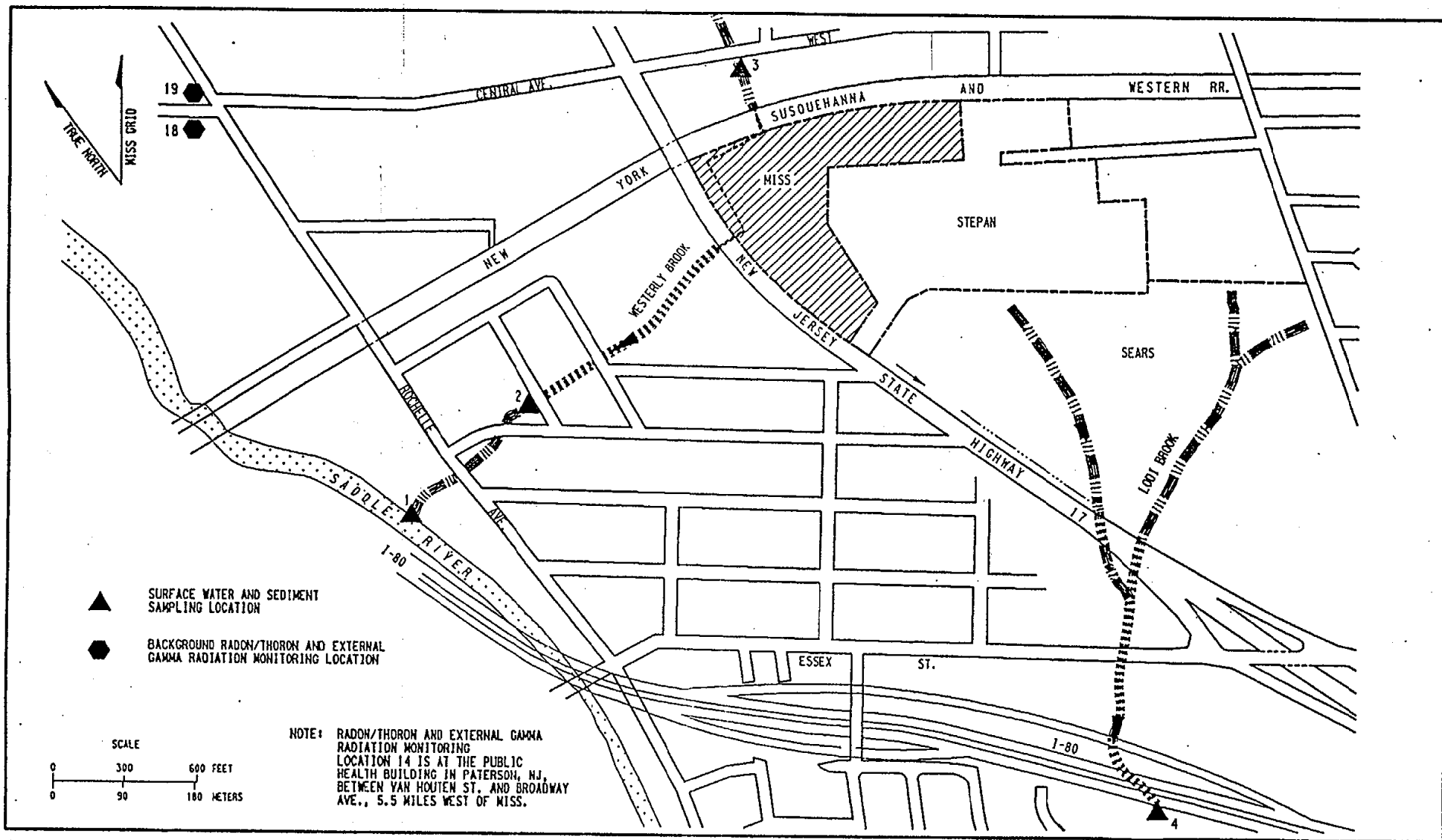
Onsite Radon, Thoron, and External Gamma Radiation Monitoring Locations

109431



Groundwater Monitoring Well Locations





138PK1A2.DGN

Offsite Radon/Thoron, External Gamma Radiation, Surface Water, and Sediment Monitoring Locations in the MISS Area

109431